Amidrazones. 6 (1). Synthesis of 1,2,4-Thiadiazoles by Thermolysis of N^3 -Thiocarbamoylamidrazone Ylides

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Reaction of N^3 -unsubstituted amidrazone ylides (**1a** and **1b**) with alkyl or aryl isothiocyanates gives N^3 -thiocarbamoylamidrazone ylides (**2**). Thermolysis of **2** gives 3-alkyl(or aryl)-5-alkyl(or aryl)amino-1,2,4-thiadiazoles (**3a-i**).

I. Heterocyclic Chem., 18, 201 (1981).

In paper 4 (2) of this series, we reported that N^3 -unsubstituted amidrazone ylides (1) are readily obtained by the base-promoted reaction of a nitrile with 1,1,1-trimethyl-hydrazinium chloride (or tosylate).

In this paper, we report that alkyl or aryl isothiocyanates add to **1a** and **1b** to give excellent yields of N^3 thiocarbamoyl substituted amidrazone ylides (**2a-i**) (3). When heated at their decomposition temperatures, **2a-i** undergo intramolecular nucleophilic attack by sulfur at N^2 to give trimethylamine and good yields of 3-alkyl(or aryl)-5-alkyl(or aryl)amino-1,2,4-thiadiazoles (**3a-i**).

We have also found that p-tolyl isocyanate reacts with 1a to give the N^3 -carbamoyl substituted ylide 4. Thermolysis of 4 afforded a complex mixture of products from which only 4,4'-dimethylcarbanilide could be isolated. In view of the unpromising results obtained from the thermolysis of 4, we did not prepare further examples of N^3 -carbamoyl substituted ylides.

In two cases, we established that good yields of thiadiazoles (3a and 3c) may be obtained by a "one-flask" procedure that does not involve isolation and purification of the intermediate ylides (2). Although this method was not applied to the synthesis of the other thiadiazoles in the series, it should be generally applicable.

We regard our procedure as a facile method for the synthesis of the 1,2,4-thiadiazoles of type 3. Previously employed methods for the preparation of 3 have been summarized by Kurzer and Tertiuk (4).

EXPERIMENTAL

Melting points were determined using a Mel-temp apparatus and are uncorrected. Nmr spectra were determined with a Hitachi-Perkin Elmer R24B spectrometer utilizing hexamethyldisiloxane as an internal standard. Infrared spectra were determined with a Perkin-Elmer 710B instrument. Yields are not maximized.

Preparation of N3-Thiocarbamoylamidrazone Ylides (2).

To 0.01 mole of the isothiocyanate in 20 ml. of dry benzene was added 0.01 mole of 2-(iminophenylmethyl)-1,1,1-trimethylhydrazinium hydroxide inner salt (1a) or 2-(1-iminoethyl)-1,1,1-trimethylhydrazinium hydroxide inner salt (1b). The reaction mixture was warmed on the steam bath to dissolve the added ylide. Immediate separation of the crystalline products was usually observed.

Preparation of 3-Alkyl(or aryl)-5-alkyl(or aryl)amino-1,2,4-thiadiazoles (3).

When heated at the decompositon temperatures given in Table I, 2a-i underwent vigorous evolution of trimethylamine. The crude products were purified by recrystallization. With the exception of 3b, the products have been previously reported (4,5,6) and in all cases the nmr spectra were compatible with the assigned structure. Compounds 3a and 3c were prepared by the published procedures and were found to be identical (nmr, ir) with the products obtained from 2a and 2c.

"One-Flask" Preparation of Thiadiazoles 3a and 3c.

Phenyl isothiocyanate (0.24 g.) was mixed with **1a** (0.31 g.). After the initial exothermic reaction had subsided, the contents of the reaction flask were heated at 210° for 5 minutes. Recrystallization of the crude product from methanol gave **3a**, 0.29 g. (66%), m.p. 168-171°.

In a similar fashion (cyclization temperature 175°), **3c** was obtained in 62% yield, m.p. 139-145°.

Table I
Analytical and Spectroscopic Data (2a-i)

Compound No.	M.p. °C	Yield (a)	Molecular Formula	Analysis Calcd. (Found)			Ir (cm ⁻¹) (b)		Nmr δ (c)
				С	Н	N	NH	C=S	
2a	188-189 (d)	77	$C_{17}H_{20}N_{4}S$	65.35 (65.16)	6.45 (6.48)	17.93 (17.88)	3225	1240	3.4 (s, 9H), 7.0-7.7 (m, 10H), 9.1 (s, 1H)
2 b	173-174 (d)	100	$C_{21}H_{22}N_4S$	69.58 (69.36)	6.12 (6.18)	15.46 (15.41)	3250	1230	3.4 (s, 9H), 7.1-7.7 (m, 12H), 9.2 (s, 1H)
2 c	164-165 (e)	61	$C_{12}H_{18}N_{4}S$	57.57 (57.61)	7.25 (7.29)	22.38 (22.35)	3240	1230	2.8 [d, 3H, (s on NH → ND)], 3.4 (s, 9H), 6.9-7.5 (m, 6H, NH and aromatic)
2 d	164-165 (f)	64	$C_{13}H_{20}N_4S$	59.05 (59.15)	7.63 (7.63)	21.19 (21.14)	3250	1220	(deuteriochloroform), 1.2 (t, 3H), 3.1-3.6 [s (3.4) superimposed on m (CH ₂ CH ₃), 11H], 5.9 (bd, 1H), 7.1-7.8 (m, 5H)
2 e	166-167 (f)	50	$C_{17}H_{26}N_4S$	64.11 (64.24)	8.23 (8.26)	17.59 (17.54)	3250	1220	0.9-2.0 (m, 10H), 3.4 (s, superimposed on m, 10H), 6.7 (bd, 1H), 7.1-7.5 (m, 5H)
2f	159-160 (f)	65	$C_{14}H_{20}N_4S$	60.83 (60.78)	7.29 (7.30)	20.27 (20.21)	3220	1230	3.3 (s, 9H), 3.7-4.0 (m, 2H), 4.8-5.3 (m, 2H), 6.5-6.9 (m, 1H), 6.7-7.6 (m, 6H, NH and aromatic)
2g	150-152 (e)	83	$C_{12}H_{18}N_{4}S$	57.57 (57.63)	7.15 (7.28)	22.38 (22.31)	3220	1230	2.1 (s, 3H), 3.2 (s, 9H), 6.7-7.4 (m, 5H), 9.0 (s, 1H)
2h	147-149 (f)	87	$C_7H_{16}N_4S$	44.64 (44.60)	8.57 (9.61)	29.75 (29.67)	3220	1220	(deuteriochloroform), 2.2 (s, 3H), 2.8 (d, 3H), 3.3 (s, 9H), 6.1 (bd, 1H)
2i	153-155 (f)	62	$C_{12}H_{24}N_4S$	56.20 (56.24)	9.43 (9.46)	21.85 (21.83)	3240	1210	(deuteriochloroform), 1.0-2.0 (m, 11H), 2.2 (s, 3H), 3.4 (s, 9H), 6.0 (bd, 1H)

⁽a) Yield of crude product. (b) Potassium bromide pellets. (c) Unless otherwise specified, determined in DMSO-d₆. (d) Recrystallized from N,N-dimethylformamide. (e) Recrystallized from methanol. (f) Recrystallized from ethanol.

Table II

Properties of 1,2,4-Thiadiazoles (3a-i)

Compound	Yield (a)	M.p. °C	Literature M.p. °C	Reference	Molecular Formula	Nmr (δ) Data (b)
3a	99 (c)	164-167	173-174	(4)	C14H11N3S	6.8-8.3 (m, 10H), 11.1 (s, 1H, exchangeable)
3b	72	143-145	(d)		$C_{18}H_{14}N_{3}S$	7.3-8.3 (m, aromatic and NH)
3 c	73	155-157	155-157	(4)	C,H,N,S	(deuteriochloroform), 2.9 [d, 3H (s on NH - ND)], 7.2-7.6 (m, 4H, aromatic and NH), 8.0-8.4 (m, 2H)
3d	65	73-75	75	(5)	$C_{10}H_{11}N_3S$	1.2 (t, 3H), 3.3 [m, 2H (t on NH → ND)], 7.1-7.6 (m, 3H), 7.8-8.2 (m, 2H), 8.4 (bd, 1H, exchangeable)
3 e	90	120-122	129	(5)	$C_{14}H_{17}N_3S$	1.1-2.1 (m, 10H), 3.4 (m, 1H), 7.2-7.5 (m, 3H), 7.9-8.2 (m, 2H), 8.4 (d, 1H, exchangeable)
3f	79 (e)	76-78	82	(6)	$C_{11}H_{11}N_3S$	4.0 (m, 2H), 5.2 (m, 2H), 5.7-6.1 (m, 1H), 7.3-7.6 (m, 3H), 7.8-8.2 (m, 2H), 6.8 (bd, 1H, exchangeable)
3 g	74	106-109	113-114	(4)	C,H,N,S	2.3 (s, 3H), 6.9-7.5 (m, 5H), 10.2 (bd, 1H, exchangeable)
3h	67	90-92	96	(5)	C ₄ H ₇ N ₃ S	(deuteriochloroform), 2.4 (s, 3H), 3.0 (s, 3H), 8.1 (bd, 1H, exchangeable)
3i	48	97-99	103	(5)	C ₉ H ₁₅ N ₃ S	(deuteriochloroform), 0.8-2.5 [s (2.4) superimposed on m, 13H], 3.1 (m, 1H), 6.8 (bd, 1H, exchangeable)

⁽a) Yield of recrystallized product. Recrystallization solvents utilized were those specified in the literature. (b) Unless otherwise specified, spectra were determined in DMSO-d₆. (c) Crude, unrecrystallized material that was identical (nmr, ir) with an authentic sample. (d) Anal. Calcd.: C, 71.26; H, 4.32; N, 13.85. Found: C, 71.09; H, 4.38; N, 13.80. (e) Recrystallized from aqueous ethanol.

Preparation and Thermolysis of 1,1,1-Trimethyl-2-[phenyl[[(4-methyl-phenylamino]carbonyl]amino]methylene]hydrazinium Hydroxide Inner Salt (4)

Addition of p-tolylisocyanate (1.33 g.) to 1a (1.77 g.) resulted in an exothermic reaction. The crude solid was pulverized under 20 ml. of dry benzene. Filtration afforded 4 as a white solid, 2.1 g., m.p. 194-197° dec. Recrystallization from ethanol gave white crystals, m.p. 204-205° dec.; nmr (deuteriochloroform): d 2.1 (s, 3H), 3.3 (s, 9H), 6.6 (s, bd, exchangeable), 6.8-7.5 (m, 9H); ir (potassium bromide): 3280 and 1620 cm⁻¹.

Anal. Calcd. for C₁₈H₂₂N₄O: C, 69.65; H, 7.15; N, 18.05. Found: C, 69.75; H, 7.20; N, 17.99.

Compound 4 (1.0 g.) was heated for 15 minutes at 210°. The decomposition was accompanied by the vigorous evolution of a basic gas. Tlc of the dark red mixture (silica gel 1:20 ethanol/chloroform) revealed 8 components. Treatment of the crude mixture with 10 ml. of boiling benzene gave 80 mg. of 4.4'-dimethylcarbanilide, m.p. 253-257°, lit. m.p. 264° (7), nmr (DMSO- d_6): δ 2.2 (s, 6H), 7.0 and 7.3 (aromatic AB, 8H), 8.5 (s, 2H). The nmr spectrum of the product was identical with that obtained from an authentic sample.

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